## Title: Self-assembly behavior of functional polycaprolactones

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## Abstract

Amphiphilic homo polycaprolactones (PCLs) having di-, tri-, tetra(ethylene glycol) (ME<sub>2</sub>, ME<sub>3</sub>, ME<sub>4</sub>) mono functional groups and a tri(ethylene glycol)/dodecyl (ME<sub>3</sub>DD) di functional group were synthesized by ring opening polymerization (ROP) of the corresponding  $\gamma$ -functionalized  $\varepsilon$ -caprolactone (CL) monomers. All the homopolymers formed spherical micelles in aqueous media with variable sizes between 73.2 nm-152.0 nm. The critical micelle concentration (CMC) for PME<sub>2</sub>CL, PME<sub>3</sub>CL, PME<sub>4</sub>CL, and PME<sub>3</sub>DDCL were  $2.2 \times 10^{-1}$ ,  $2.4 \times 10^{-1}$ ,  $3.7 \times 10^{-1}$  gL<sup>-1</sup>, and  $1.8 \times 10^{-2}$  gL<sup>-1</sup>, respectively. With the increasing of hydrophilic oligo(ethylene glyco) chain length, the polymeric micelles became less stable and their cloud point tempeature increased from 58.6 °C-85.8 °C. PME<sub>3</sub>DDCL micelles were more stable due to the increased hydrophobicity from the additonal dodecyl functional group and had the highest loading capacity of 4.8%. However, it significantly affected the cloud point tempeature  $(T_{cp})$  of PME<sub>3</sub>DDCL which was not detected even lowing temperature to 4 °C. Compared with homopolymers, amphiphilic blcok copolymers tend to have higher thermodynamic stability. Therefore, a new amphiphilic diblcok copolymer polycaprolatone-block-poly(triethylene glyco/propyl-disubstituted caprolacotne) (PCL<sub>44</sub>-*b*-P(ME<sub>3</sub>PyCL)<sub>56</sub>) with a mocelular weight of 5,300 g/mol and a PDI of 1.69 was preparated by sequential polymerization. The self-assembly was achieved in aqueous media and the CMC was  $2.41 \times 10^{-3}$  gL<sup>-1</sup>. It is comparable to that  $(1.54 \times 10^{-3} \text{ gL}^{-1})$  of polycaprolactone-block-poly(triethylene glycol) (PCL<sub>48</sub>-*b*-P(ME<sub>3</sub>)<sub>52</sub>) diblock copolymer with a molecular weight of 6,000 g/mol and a PDI of 1.38.

